An Ambipolar BODIPY Derivative for a White Exciplex OLED and Cholesteric Liquid Crystal Laser toward Multifunctional Devices

Marian Chapran,†‡ Enrico Angioni,§ Neil J. Findlay,§ Benjamin Breig,§ Vladyslav Cherpak,⊥ Pavlo Stakhira,⊥ Tell Tuttle,§ Dmytro Volyniuk,‖ Juozas V. Grazulevicius,⊥ Yuriy A. Nastishin,⊥ Oleg D. Lavrentovich,*‡⊥ and Peter J. Skabara*§⊥

†Department of Molecular Physics, Lodz University of Technology, Zeromskiego 116, 90-924 Lodz, Poland
‡Lviv Polytechnic National University, S. Bandera 12, 79013 Lviv, Ukraine
§WestCHEM, Department of Pure and Applied Chemistry, University of Strathclyde, Glasgow G1 1XL, U.K.
‖Department of Polymer Chemistry and Technology, Kaunas University of Technology, Radvilenu Plentas 19, LT-50254 Kaunas, Lithuania
⊥Liquid Crystal Institute and Chemical Physics Interdisciplinary Program, Kent State University, Kent, Ohio 44242, United States

ABSTRACT: A new interface engineering method is demonstrated for the preparation of an efficient white organic light-emitting diode (WOLED) by embedding an ultrathin layer of the novel ambipolar red emissive compound 4,4-difluoro-2,6-di(4-hexylthiopen-2-yl)-1,3,5,7,8-pentamethyl-4-bora-3a,4a-diaza-indacene (bThBODIPY) in the exciplex formation region. The compound shows a hole and electron mobility of $3.3 \times 10^{-4}$ and $2 \times 10^{-4}$ cm$^2$ V$^{-1}$ s$^{-1}$, respectively, at electric fields higher than $5.3 \times 10^{5}$ V cm$^{-1}$. The resulting WOLED exhibited a maximum luminance of 6579 cd m$^{-2}$ with CIE 1931 color coordinates (0.39; 0.35). The bThBODIPY dye is also demonstrated to be an effective laser dye for a cholesteric liquid crystal (ChLC) laser. New construction of the ChLC laser, by which a flat capillary with an optically isotropic dye solution is sandwiched between two dye-free ChLC cells, provides photonic lasing at a wavelength well matched with that of a dye-doped planar ChLC cell.

KEYWORDS: organic light-emitting diodes (OLEDs), white light, exciplex, BODIPY dye, cholesteric liquid crystal (ChLC) laser

INTRODUCTION

Recently, white organic light-emitting diodes (WOLEDs) have attracted the attention of the scientific community for their promising application in lighting technologies.1–3 The fundamental parameters that determine the performance of WOLEDs in display technologies and lighting systems are the power efficiency, color chromaticity, and the cost of manufacturing. Already there are competitive technologies that have enabled successful improvement of these parameters, such as phosphorescent WOLEDs, which have dominated in terms of efficiency.2,4 In this case, the efficiency of WOLEDs is based on the use of guest (phosphorescent metal–organic complexes) and host materials4 but can be associated with some technical challenges including issues determining an appropriate concentration of the guest in the host system and the engineering of appropriate multilayer sandwich device configurations.4,5 Moreover, the host and phosphorescent materials should be designed with suitable triplet and singlet energy levels, especially considering the energies of the blue luminescent component.3 In 2013, Zhao et al.6 proposed an alternative device architecture that utilized an ultrathin (0.1 nm) phosphorescent nondoped emissive layer for the formation of a highly efficient phosphorescent WOLED. However, such an approach does not provide a satisfactory efficiency roll-off which is especially important for the development of attractive lighting systems.7

One of the most simple and conceptual approaches for the development of WOLEDs is the introduction of yellow emission generated by an exciplex (bimolecular excited species). This emission can be used in combination with blue emission from a fluorescent excited state$8–10$ or from a second exciplex state, as described recently by Chou and co-workers.11 Exciplex-type OLEDs are characterized by high values of external quantum efficiency due to the intramolecular spin up-conversion from the nonradiative triplet excited state to the singlet excited state (thermally activated delayed fluorescence (TADF)).11,12 Despite the high efficiency of all-exciplex-based OLEDs, the main limiting factor in the development of a real
white OLED is the limited number of known materials that can form exciplexes with emission in the blue and red regions of the visible spectrum. Recently, exciplexes have been used as hosts for highly efficient yellow and red phosphorescent OLEDs as well as for blue phosphorescent OLEDs. The energy transfer from the exciplexes to the dopants was efficient with no significant energy loss. Long-range exciplex emission was observed recently in spatially separated electron-donating and electron-accepting molecules across an ultrathin spacer layer.

Here we present a novel red emitter, 4,4-difluoro-2,6-di(4-hexylthiophen-2-yl)-1,3,5,7,8-pentamethyl-4-bora-3a,4a-diaza-s-indacene (bThBODIPY), that was synthesized using an adapted literature procedure. 4,4-Difluoro-4-bora-3a,4a-diaza-s-indacene (BODIPY) derivatives are characterized by small Stokes shifts, high thermal and photochemical stabilities, high photoluminescence quantum yields (PLQYs), good solubility, and general chemical robustness. They have been used as organic semiconductors for different applications, including OLEDs, OPFs, OFETs, and lasers, as well as finding applications as biochemical labels, fluorescent switches, and chemosensors.

In bThBODIPY, the thiophene substituents are electron-rich and the BODIPY unit is electron-deficient, resulting in a donor−acceptor structure. Alkyl chains were introduced with the aim to increase the solubility of the new material in common organic solvents (and hence their processability) and to avoid intermolecular packing. In fact, BODIPY molecules, due to their planarity, commonly exhibit “aggregation-caused quenching (ACQ)” effects: showing strong fluorescence in dilute solution but weaker luminescence in the solid state. In such a molecule, the efficiency of the energy transfer through the BODIPY core is generally increased, producing an elevated PLQY as demonstrated in similar β-substituted BODIPY molecules. Here, we present a new interface engineering method for the preparation of efficient white OLEDs by embedding an ultrathin layer of the new highly efficient fluorescent red emitter bThBODIPY in the exciplex formation region. The resulting WOLED presented a combined emission from different excited states: red exciton emission from the ultrathin bThBODIPY film, green emission from the m-MTDATA (4,4′,4′′-tris[phenyl-(m-tolyl)amine]-triphenylamine)/ PPT (2,8-bis(diphenylphosphoryl)dibenzof[b,d]thiophene) exciplex, and blue emission from m-MTDATA excitons and excimer. A prerequisite for the implementation of this approach is a theoretical model for the electronic processes at the organic exciplex interfaces, which shows that the intermolecular excited state may form even if the donor and acceptor layers are separated.

The compounds m-MTDATA and PPT have been employed here in order to obtain a green exciplex based emission. This strategy has been previously applied in the fabrication of different OLEDs and demonstrated one of the highest external quantum yields (10%) reported. The outstanding fluorescence characteristics of the dye bThBODIPY can also be exploited for laser generation in cholesteric liquid crystals (ChLC). Two experimental approaches for optically pumped mirrorless lasing in a dye-doped ChLC have been tested. In the first approach, laser generation was examined in a conventional geometry, where the dye-doped ChLC was filled into a flat cell with tangential alignment of the ChLC molecules at (helical axis normal to) the cell substrates (from herein referred to as the planar cell). In the second approach, a flat microcapillary filled with a toluene solution of bThBODIPY was placed between two flat planar dye-free ChLC cells. Toluene was chosen as a solvent since it evaporates relatively slowly in comparison with other solvents, and this is known to provide good quality, uniform films. Stable optically pumped lasing was achieved with both approaches.

**Results and Discussion**

**Synthesis.** The emissive bThBODIPY material was obtained in 34% yield by Stille coupling between 4,4-difluoro-2,6-diiodo-1,3,5,7,8-pentamethyl-4-bora-3a,4a-diaza-s-indacene and (4-hexylthiophen-2-yl)trimethylstannane using tetrakis(triphenylphosphine)palladium(0) as the catalyst (Figure 1).

**Figure 1.** Synthesis of bThBODIPY.

![Synthesis of bThBODIPY](image)

The absorption and emission spectra of the solution (10⁻⁵ M in CH₂Cl₂) and of the vacuum-deposited film of bThBODIPY were recorded (Figure 2). The absorption spectrum of the solution showed four bands with maxima at 524, 394, 277, and

**Figure 2.** (a) Experimental UV−vis absorption spectra in solution (10⁻⁵ M in CH₂Cl₂, black) and for the vacuum deposited film (gray) of bThBODIPY, calculated vertical absorptions (red) and Gaussian fitted (fwhm = 0.37 eV) absorption spectra at the TD-DFT/PBE0/6-311G(d,p) level of theory (dashed red) on the vacuum-deposited film of bThBODIPY. (b) Photoluminescence spectra in solution (10⁻⁵ M in CH₂Cl₂, black) and for the vacuum deposited film of bThBODIPY.
241 nm, while the solid sample exhibited maxima at 516, 387, 275, and 244 nm. The slight blue-shift of the film spectrum is due to the general larger population of the less energetic vibrational states due to the smaller reorganization energy in the solid state. However, the edge of the lowest energy absorption band of the film showed a red-shift when compared with that of the solution. The sharp profile of the less energetic band is characteristic of the BODIPY unit of the molecule, whereas the other bands have charge transfer character between the donor and the acceptor components, as observed for similar BODIPY derivatives. From the longest wavelength absorption edge of the UV–vis absorption spectra in solution, it was possible to calculate an optical HOMO–LUMO gap of 2.2 eV. The fluorescence spectra showed a main band, with a large Stokes shift, centered at 605 and 609 nm for the solution and the film of bThBODIPY, respectively. In the spectrum of the film, the main emission was associated with a shoulder centered at ca. 636 nm, which is likely due to intermolecular interactions between the bThBODIPY molecules that are more likely to occur in the solid state. Fluorescence decay measurements (recorded at 605 nm) were performed for a vacuum-deposited film of bThBODIPY. A single-exponential decay profile was observed, resulting in a lifetime of 2.21 ns that can be assigned to pure fluorescence emission. The decay spectrum recorded at 650 nm shows a fluorescence emission with a decay time of 0.92 ns, in agreement with the emission arising from the π–π stacking in the film (Figure S4). Absolute photoluminescence quantum yields (PLQYs) of 97% and 17% were obtained for the dichloromethane solution and for the thin film (vacuum deposition at 10^{-6} Torr) of bThBODIPY, respectively.

To explain the experimentally observed electronic spectra (and the chemical behavior), quantum chemical calculations were performed for bThBODIPY using density functional theory (DFT) and its time-resolved counterpart (TD-DFT). All the calculations were performed with the software package Gaussian09 (Revision A.02), with the hexyl chains replaced with methyl groups to limit the computational cost without compromising the integrity of the calculations. The geometry of bThBODIPY was optimized at the B3LYP/6-311G(d,p) level of theory for bThBODIPY. The other significant transitions that correspond to the band centered at 4.51 and 5.08 eV in the solid state (277 and 241 nm in solution) are mainly represented by vertical transitions at 4.46 and 5.17 eV. They are between orbitals delocalized over the entire molecule (HOMO, HOMO−1, HOMO−2) and orbitals mainly localized on the thiophene substituents (LUMO +1, LUMO+2). This is in good agreement with the donor–acceptor character desired for bThBODIPY and confirmed experimentally by the electrochemical data; i.e., the thiophene substituents act as donors and the BODIPY center as an acceptor. The calculated energies are generally in good agreement with the experimental data, in particular with the absorption data obtained with the thin film where there are no additional solvent effects.

The cyclic voltammograms of bThBODIPY (Figure 4) showed a reversible reduction process (E_{1/2} = −1.63 V) attributable to the typical formation of a radical anion on the BODIPY unit. The oxidation waves showed a quasi-reversible process at E_{1/2} = +0.60 V. These are due to the electron donor character of the 4-hexylthiophen-2-yl units. The electrochemical HOMO and LUMO levels were calculated from the onset of the first oxidation wave (−5.5 eV) and reduction wave (−3.2 eV), respectively (the data were referenced to ferrocene, which has a HOMO of −4.8 eV). The electrochemical HOMO–LUMO gap was therefore determined to be 2.3 eV.

**Solid-State Photoelectrical Properties of bThBODIPY.** Since solid-state layers of bThBODIPY are required for the fabrication of OLEDs, an ionization potential (IP) of 5.68 eV was measured by electron photoemission experiments on the vacuum-deposited film of bThBODIPY (Figure S6). The IP is in good agreement with the result obtained by DFT calculations (HOMO\text{DFT} = −5.6 eV) and from the electrochemical experiments (HOMO\text{EL} = −5.5 eV). The slight disagreement with the HOMO\text{EL} is likely due to intermolecular interactions (stacking) of bThBODIPY that can occur in the...
Electrons were determined demonstrating the ambipolar nature of the material and allowing the estimation of the hole and electron mobilities of bThBODIPY. The electric field dependences of hole and electron mobilities showed linear dependencies on the square root of the electric field ($E^{1/2}$). This observation is in agreement with the Poole–Frenkel relationship $\mu = \mu_0 \times \exp(\alpha \times E^{1/2})$ (Figure 5). Hole and electron mobilities of the vacuum-deposited layer of bThBODIPY were found to be $3.3 \times 10^{-4}$ and $2.0 \times 10^{-4}$ cm$^2$ V$^{-1}$ s$^{-1}$, respectively, at electric fields higher than $5.3 \times 10^5$ V cm$^{-1}$ (Figure 5). Such moderate values of charge mobilities are comparable to many light-emitting materials used in OLEDs.

The hole and electron mobilities of bThBODIPY are well balanced, showing that this new compound can be used as an emitting layer in OLEDs.

**Device Fabrication.** Two devices were fabricated by successive vacuum deposition of organic semiconductor layers and metal electrodes onto precleaned indium tin oxide (ITO)-coated glass substrates under a vacuum of $10^{-5}$ Torr. The structures of the devices were (A) ITO/CuI(8 nm)/m-MTDATA(20 nm)/bThBODIPY(70 nm)/PPT(25 nm)/Ca(50 nm)/Al(200 nm) and (B) ITO/CuI(8 nm)/m-MTDATA(100 nm)/bThBODIPY(4 nm)/PPT(60 nm)/Ca(50 nm)/Al(200 nm) (Figure S7). The active area of the WOLEDs was $3 \times 2$ mm$^2$, and the characteristics of the devices were obtained immediately after fabrication without further passivation in air. CuI was used as the hole-transporting layer. The laser generation was tested in an assembled cell with a gap of 0.1 mm and output at the repetition rate of approximately 100 Hz. The lasing emission was collected in the direction perpendicular to the substrates. The emission spectra were measured using a fiber-based spectrometer (Ocean Optics USB2000). A Molecron OPTIMUM 4001 laser energy meter was used for measuring the lasing threshold. For the capillary laser construction, a 0.3 wt% solution of bThBODIPY dye in toluene was placed into a rectangular 2 $\times$ 0.1 mm$^2$ VitroCom glass capillary, the ends of which were sealed with epoxy resin. The capillary was placed between two identical cholesteric (27.6 wt% of S811 in ZLI-2248) cells of 20 $\mu$m gap formed by glass substrates covered with a PI-2555 layer for a planar alignment. The capillary was covered with a thin layer of optical glue Norland 64 for refractive index matching and tightly assembled between the ChLC cells. The whole construction could be easily assembled/disassembled with replaceable ChLC cells and capillaries.

**OLED Characterization.** The varying thicknesses of m-MTDATA (hole-transport/hole-injection layer) and PPT (electron-transport/hole-blocking layer) are associated with different roles for each of the materials in the different OLEDs. In device A, they provide only transport and injection of carriers to the emissive layer (bThBODIPY), while in device B they additionally take part in the formation of an exciplex state. The two devices differed only by the thickness of the bThBODIPY layer (70 nm in device A and 4 nm in device B). In device A the thickness of the bThBODIPY layer (70 nm) allowed only the emission from the exciton excited state of bThBODIPY, resulting in a strong electroluminescence band at 602 nm (Figure 6). This is due to the HOMO (HO-

**MO$_{m$-MTDATA} is close to HOMO$_{bThBODIPY}$) and LUMO (LUMO$_{PPT}$ is close to LUMO$_{bThBODIPY}$) levels of the neighboring layers that facilitated the injection of holes and electrons in the emissive layer (Figure S7). In device B, the reduced thickness of the bThBODIPY layer (4 nm) allowed exciplex formation between the layers of m-MTDATA and PPT. The thickness of the emissive layer was less than the characteristic length associated with the exciplex wave function. This allowed the formation of a strong bound state between m-MTDATA and PPT layers with tunneling through the bThBODIPY spacer layer. Consequently, the EL spectrum of device B had a more complex shape and contained three additional bands ($\lambda_{max} = 428, 495, \text{and} 511 \text{ nm}$) (Figure 6). The band at 428 nm and the shoulder at 495 nm are due to the
recombination of excitons and eximers in the m-MTDATA layer,\textsuperscript{36} while the band centered at 511 nm is due to the well-known exciplex emission of the m-MTDATA:PPT couple.\textsuperscript{10} The main characteristics of devices A and B are summarized in Table 1. Device A presented CIE 1931 coordinates (0.49, 0.42), Table 1. Characteristics of Devices A and B

<table>
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<th>device A</th>
<th>device B</th>
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<tr>
<td>$V_{on}$ (V)</td>
<td>4</td>
<td>3</td>
</tr>
<tr>
<td>max brightness (cd m\textsuperscript{-2})</td>
<td>28829</td>
<td>6579</td>
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<tr>
<td>max current efficiency (cd A\textsuperscript{-1})</td>
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<td>2.26</td>
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<td>max power efficiency (lm W\textsuperscript{-1})</td>
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<td>1.2</td>
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<tr>
<td>max external quantum efficiency (%)</td>
<td>6.8</td>
<td>1.17</td>
</tr>
<tr>
<td>CIE 1931 coordinates ($x$, $y$)</td>
<td>(0.49, 0.42)</td>
<td>(0.39, 0.35)</td>
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Figure 7. Current density–voltage and brightness–voltage (a) and current efficiency–current density (b) characteristics of devices A and B.

with high brightness (28829 cd m\textsuperscript{-2}) and efficiencies (Figure 7, Table 1, and Figure S8). These are due to the relatively high emission quantum yield of bThBODIPY in the solid state in combination with the efficient recombination of holes and electrons in the transport layers; this provides the background for an excellent balance between the electrons and holes in the bThBODIPY layer. A brightness of 6579 cd m\textsuperscript{-2} at 15 V with a current efficiency of 2.26 cd A\textsuperscript{-1} was recorded for device B. The CIE 1931 coordinates of (0.39, 0.35) are close to those of the calibrated white light (0.33, 0.33). The lower operating voltage of device B, when compared with device A, is due to the barrier-free exciplex emission\textsuperscript{36} generated by electron tunneling from the PPT to the m-MTDATA layers.\textsuperscript{36} At first glance device B seems far behind in external quantum efficiency (1.17\%) (Table 1 and Figure 7) in comparison with the pure exciplex OLED based on PPT and m-MTDATA (10\%).\textsuperscript{36} However, in the pure exciplex device the efficiency drops to 0.4\% at high brightness due to the efficiency roll-off. In contrast, device B is characterized by stable external quantum efficiency over a wide range of working current densities (30–315 mA cm\textsuperscript{-2}, Figure 7). The increased separation distance between the electrons and holes in the intermolecular state (PPT:m-MTDATA exciplex) by introduction of an ultrathin layer of bThBODIPY results in a reduction of the spin exchange energy of the exciplex state ($\Delta_{CT}$) and therefore reduces the probability of triplet-polaron annihilation.\textsuperscript{13,50} Thus, the roll-off efficiency at high current densities was improved, decreasing the $\Delta_{CT}$ and broadening the exciton formation width.\textsuperscript{50}

High external quantum efficiency of the nonexciplex fluorescent device A can be explained by the advanced confinement of the hole and electron injection layers with the ambipolar bThBODIPY layer. Here, effective double injection and blocking of holes and electrons occur for the complete absence of barriers for both charge carriers at the m-MTDATA/bThBODIPY and bThBODIPY/PPT interfaces and for the high blocking barriers (1.2 eV) for both charge carriers. Another important advantage of device A is the m-MTDATA and PPT exciton confinement with bThBODIPY\textsuperscript{51} since both m-MTDATA and PPT have a larger energy gap (3.1 eV for m-MTDATA and 3.7 eV for PPT) than bThBODIPY (2.3 eV) (Figure S7). As a result, the recombination zone is broadly distributed in the ambipolar bThBODIPY layer without exciton quenching of luminescence at the electrodes.\textsuperscript{52}

bThBODIPY as a Dye for Optically Pumped Lasing in a Cholesteric Liquid Crystal. The fluorescence characteristics of bThBODIPY (Figure 2) suggest that it can be used for laser generation. There are many different approaches for the construction of dye lasers although all are based on the common principle of optical feedback resonant amplification. In commercially produced lasers the optical feedback is usually achieved with mirror resonators. Modern laser resonators are produced with high quality optical elements and require precision alignment, resulting in a bottleneck in production. Recently, significant effort has been focused on developing lasers utilizing mirrorless principles of optical feedback.\textsuperscript{53,54} One of these mirrorless feedback mechanisms is selective light reflection in cholesteric liquid crystals (ChLC),\textsuperscript{55} which under special conditions act as 1D photonic crystals.\textsuperscript{56} Light emission from a fluorescent dye dissolved in cholesteric liquid crystals is inhibited within the spectral range of the photonic band gap (PhBG),\textsuperscript{57} which is the selective reflection band for a ChLC, but leaks as strongly amplified laser radiation\textsuperscript{58,59} at the PhBG edges. As such, highly effective fluorescent dyes for ChLC lasers are in great demand. On the other hand, modern trends to develop multiple-in-one microdevices set requirements for multifunctional active materials. In this context, the newly synthesized bThBODIPY dye, demonstrated above to be an effective material for WOLEDs, can be effectively used as a dye for a ChLC laser, possibly setting the basis for the development of compact and on-chip WOLED-pumped ChLC lasers.

We used two different experimental settings to achieve laser emission, which differ in the way the bThBODIPY dye (acting as the active element) and the ChLC (serving to provide the optical feedback) are combined. A first laser was obtained following a classical approach,\textsuperscript{55} in which the dye was dissolved in a planar layer of ChLC. In the second setting, the dye was dissolved in an optically isotropic solvent; a flat capillary with this isotropic solution was sandwiched between two identical dye-free ChLC planar cells, serving as Bragg mirrors of resonance. The second setting, which to the best of our
knowledge has never been reported before, will be referred herein after as “a sandwich capillary laser”.

In order to have efficient laser generation, the cholesteric PhBG must be overlapped with the dye fluorescence emission band (DFEB). The spectral location of the PhBG of a cholesteric mixture is defined by the concentration of the chiral dopant in a liquid crystalline matrix. Figure 8 shows an overlap of the cholesteric PhBG with the DFEB. Laser generation produced by the first setting, with the dye-doped cholesteric cell (Figure 9a and Figure S9), was registered (blue line in Figure 8) at the long-wavelength edge of the cholesteric PhBG (black line in Figure 8), confirming its photonic nature. Lasing was stable for excitation energy values exceeding the threshold of ca. 0.6 μJ pulse⁻¹ (Figure 10).

The laser emits in both the forward and backward directions, meaning that the true output lasing energy is double the value measured with the energy meter (in the forward direction), as shown in Figure 10. The second setting, based on the sandwich capillary approach described above, produces lasing similar to the classic case of ChLC doped with the dye. Excitation of the isotropic solution of dye with a laser beam at an incidence angle of 10°, focused inside the capillary, resulted in stable lasing (Figure 9b and Figure S9) at the long-wavelength edge of the cholesteric PhBG (red line in Figure 8).

The emission was close to that obtained in the dye-doped ChLC cell (blue line in Figure 8) and confirms that the lasing effect is photonic in nature, even though the dye is not dissolved in the ChLC. Stable lasing was registered for excitation energy values above 0.9 μJ pulse⁻¹ (red line in Figure 10). This energy is considerably higher than the activation threshold for lasing in the dye-doped cell and is associated most probably with the loss of energy at the capillary interfaces. Additionally, there is a strong emission leak from the capillary ends. This means that the wave-guiding lasing modes take some part of the emission energy, thereby decreasing the output of the PhBG mode. Although the lasing energy activation threshold is higher for the capillary construction than for the conventional dye-doped cell, the capillary construction can be very convenient when the amount of emissive material is limited. Furthermore, the same capillary laser can be inserted between different pairs of ChLC cells that exhibit different PhBGs, leading to spectral shifts of the laser emission. Conversely, this new method allows for rapid screening of potential lasing materials through altering the capillary within the same pair of ChLC cells.

CONCLUSIONS

The new ambipolar red emissive compound bThBODIPY was synthesized, and the optical, photophysical, and electrochemical properties were studied. Hole and electron mobilities were estimated for bThBODIPY using the time-of-flight technique with values of 3.3 × 10⁻⁴ and 2 × 10⁻⁴ cm² V⁻¹ s⁻¹ (at electric fields higher than 5.3 × 10⁵ V), respectively. The BODIPY derivative was used, first as a single emitting layer and then as a spacer, in the fabrication of two OLEDs. A new interface engineering method for the preparation of an efficient white OLED by embedding an ultrathin layer of bThBODIPY in the exciplex formation region (between layers of PPT and m-MTDATA) was developed. The resulting OLED displayed highly efficient exciplex and exciton emission simultaneously, resulting in white emission with CIE 1931 color coordinates (0.39, 0.35) close to those of calibrated white light. The WOLED achieved stable current efficiency over a wide range of working current densities with next to no efficiency roll-off. The bThBODIPY material was further employed as a ChLC laser dye. A new construction method in which a flat capillary filled with an optically isotropic dye solution is sandwiched between two dye-free planar ChLC cells was described. This provides optically pumped lasing at the same wavelength as in the classical scheme of a dye-doped planar ChLC cell. Facile assembly/disassembly of the capillary laser can be useful for optimization of laser characteristics using small amounts of dye and ChLCs; thus, it can be used as a simple method to explore new active laser media and to optimize the concentration dependencies of the lasing parameters.

![Figure 8. Fluorescence spectrum of the bThBODIPY dye in toluene (2) and in the ChLC (5), reflection spectrum of the ChLC cell (1), laser emission spectra from the flat dye-doped ChLC cell (4), and from capillary with dye solution in toluene (3), sandwiched between two dye-free ChLC cells.](image)

![Figure 9. Photographs of laser emission patterns formed by (a) the doped ChLC cell and (b) capillary sandwich laser media.](image)

![Figure 10. Laser emission energy as a function of the excitation energy for a dye-doped cholesteric cell (black squares) and capillary with dye solution (red stars) sandwiched between two dye-free cholesteric cells.](image)
Experimental section, characterization of bTHBODIPY (1H NMR, 13C NMR, TGA, DFT calculation, photoluminescence excitation and emission spectra, and fluorescence decay curves of bTHBODIPY for a vacuum-deposited film recorded at 605 and 650 nm), energy band diagram of devices, CIE 1931 chromaticity coordinates of OLEDs, and photographs of the laser devices (PDF)

**REFERENCES**


